BINARY STARS

 $(m_1 + m_2) \propto \frac{m}{p^2}$ M_1 m_{γ}

The masses of the individual stars can be calculated.

By gathering the masses of a large variety of stars in binary systems a fundamental relationship soon became apparent.





Thermonuclear Fusion

Proton – Proton Cycle 4H ® 1He + 2g

$4H = 6.693 \times 10^{-24} \text{ gm}$ -1He = 6.645 \times 10^{-24} \text{ gm}

Difference of 4.8x10⁻²⁶ gm (0.7%)

4.8 x 10⁻²⁶ gm

E=mc

LIGHT

Some incredible numbers... The proton-proton cycle occurs 10³⁸ times/second

Each second:

624 million tons of hydrogen Fuses to become 620 million tons of helium 4 million tons of hydrogen becomes energy

$M_{\pi} = 1.99 \times 10^{30}$ kilograms

Sun's lifetime ~ 10 billion years







30% – 40% of total mass is lost

2. Intermediate Mass Stars 0.5 < M, < 8</p>













Planetary Nebula (has nothing to do with planets!!)





White Dwarf Stars

n Composed mostly of carbon n Surface temperatures of 50,000 K or more n <u>NO</u> internal energy source n Earth sized n Mass is that of remnant stellar core n VERY DENSE!

White Dwarf Star

11,000 tons per cubic inch

Limit ~ 1.4 solar M

40 Eridanus B

The electrons did!

Gravits

Electrons have a limit to how tightly they can be packed together
"ELECTRON DEGENERACY PRESSURE"

BUT! Electron Degeneracy Pressure has its limits

Gravity can overwhelm the electrons if the mass is high enough..

M < 1.4 M. Chandrasekhar Limit



What happens if the core of the star that remains is GREATER than the Chandrasekhar Limit?

3. High Mass Stars M. > 8





25 M. star <u>Element</u> Hydrogen Helium Carbon Neon Oxygen Silicon Iron

Temperature 4x10⁷ K 2x10⁸ K 6x10⁸ K 1.2x10⁹ K 1.5x10⁹ K 2.7x10⁹ K none!

Duration 7x10⁶ yrs 5x10⁵ yrs 600 yrs 1 year months days hours





Silicon \rightarrow Iron

Iron core

Iron core < 1.4M. Continual silicon fusion increases mass of core Eventually Iron core = 1.4M.

Iron core > 1.4M. Iron core cannot support itself against gravity Iron core collapses...

Fe





Supernova 1987a




















July, 1054 A.D.





Synthesis of the Elements

$\underbrace{\overset{1}{\underset{1.008}{\underline{H}}}}^{1}$	2 IIA 2A											13 IIIA 3A	14 IVA 4A	15 VA 5A	16 VIA 6A	17 VIIA 7A	2 <u>He</u> 4.003
3 <u>Li</u> 6.941	$\frac{4}{\frac{\text{Be}}{9.012}}$											5 <u>B</u> 10.81	6 <u>C</u> 12.01	7 <u>N</u> 14.01	8 0 16.00	9 <u>F</u> 19.00	10 <u>Ne</u> 20.18
11 <u>Na</u> 22.99	$\frac{12}{Mg}$ 24.31	3 IIIB 3B	4 IVB 4B	5 VB 5B	6 VIB 6B	7 VIIB 7B	8	9 VI	10 III	11 IB 1B	12 IIB 2B	$\frac{13}{\underline{Al}}_{26.98}$	14 <u>Si</u> 28.09	15 <u>P</u> 30.97	16 <u>S</u> 32.07	17 <u>C1</u> 35.45	$\frac{18}{\underline{Ar}}{}_{39.95}$
$\underbrace{\overset{19}{\underline{K}}}_{39,10}$	$\frac{\overset{20}{\underline{Ca}}}{\overset{40.08}{\underline{ca}}}$	$\frac{\frac{21}{Sc}}{\frac{44.96}{44.96}}$	22 <u>Ti</u> 47.88	23 <u>V</u> 50.94	$\frac{\overset{24}{\text{Cr}}}{\overset{52.00}{\text{52.00}}}$	25 <u>Mn</u> 54.94	26 Fe 55.85	27 <u>Co</u> 58.47	28 <u>Ni</u> 58.69	29 <u>Cu</u> 63.55	30 <u>Zn</u> 65.39	31 <u>Ga</u> 69.72	32 <u>Ge</u> 72.59	33 <u>As</u> 74.92	34 <u>Se</u> 78.96	35 <u>Br</u> 79.90	36 <u>Kr</u> 83.80
37 <u>Rb</u> 85.47	38 <u>Sr</u> 87.62	39 <u>Y</u> 88.91	$\frac{40}{Zr}$ 91.22	41 Nb 92.91	42 Mo 95.94	43 <u>Tc</u> (98)	44 <u>Ru</u> 101.1	45 <u>Rh</u> 102.9	$\frac{46}{Pd}$	47 Ag 107.9	48 <u>Cd</u> 112.4	49 <u>In</u> 114.8	50 <u>Sn</u> 118.7	51 <u>Sb</u> 121.8	52 <u>Te</u> 127.6	53 <u>I</u> 126.9	34 <u>Xe</u> 131.3
58 <u>Cs</u> 132.9	56 <u>Ba</u> 137.3	57 <u>La</u> * 138.9	72 <u>Hf</u> 178.5	73 <u>Ta</u> 180.9	$\frac{\frac{74}{W}}{183.9}$	$\frac{\frac{75}{\text{Re}}}{\frac{186.2}{186.2}}$	$\frac{\overset{76}{\text{Os}}}{\overset{190.2}{190.2}}$	$\frac{17}{10}$	78 <u>Pt</u> 195.1	79 <u>Au</u> 197.0	80 <u>Hg</u> 200.5	$\frac{81}{11}{204.4}$	$\frac{82}{Pb}_{207.2}$	83 <u>Bi</u> 209.0	84 <u>Po</u> (210)	85 <u>At</u> (210)	86 <u>Rn</u> (222)
										1.25							



Isotopes of the elements

 ${}^{12}C = 6 \text{ protons} + 6 \text{ neutrons}$ ${}^{13}C = 6 \text{ protons} + 7 \text{ neutrons}$ ${}^{14}C = 6 \text{ protons} + 8 \text{ neutrons}, \text{ unstable } t_{1/2} = ~6000 \text{ years}$ ${}^{15}C = 6 \text{ protons} + 9 \text{ neutrons}, \text{ unstable } t_{1/2} = ~12 \text{ years}$





Historical

1930's Hans Bethe discovers mechanisms by which stars shine
fusion of hydrogen to helium primary energy source

• In the 1940's and early 1950's as Big Bang picture for origin of Universe was developing – elements cooked up early in expansion

• Early 1950's this started to give way to the stars being the most likely place

Fred Hoyle, Cambridge
William Fowler, Cal Tech
Geoffrey Burbidge
Margarate Burbidge

• The seminal observation was detection of technetium in atmospheres of old (>several 10⁹ years) stars



The 3rd Dimension of the Periodic Table

Valley of Stability





Nucleosynthesis in Stars by s-process (slow neutron capture)

1	Se66	Se67	Se68	Se69	Se70	Se71	Se72	Se73	Se74	Se75	Se76	Se77	Se78	Se79	Se80
	0+	~	0+	(3/2-)	0+	3/2-5/2-	0+	9/2+	0+	5/2+	0+	2-		→	0+
		ECp	EC	ECp	EC	EC	EC	EC	0.89	EC	9.36	7.65	23.78	β.	49.61
	A\$65	As66	A367	A:68	As69	As70	As71	As/2	As73	As74	As75	A576	As77	As78	As79
			(5/2-)	3-	5/2-	4(+)	5/2-	2-	3/2-	2-	3/2	\rightarrow	32-	2	3/2-
	EC	EC	EC	EC	EC	EC	EC	EC	EC	EC,8	100	β	8	β.	β.
	Ge64	Ge65	Geőő	Ge67	Ge68	Ge69	Ge70	Ge71	Ge72	Ge73	Ge74	Ge75	Ge76	Ge77	Ge78
	0+	(3/2)-	0+	1/2-	0+	5/2-	0+	.	•) +		\rightarrow	0+	7/2+	0-
er	EC	ECp	EC	EC	EC	EC	21.23	EC	27.66	7.75	35.94	β .	7.44	9	9
p	Ga63	Ga64	Ga65	Ga66	Ga67	Ga68	Ga69	Ga70	Ga71	Ga72	Ga73	Ga74	Ga75	Ga76	Ga77
	3/2-5/2-	0+	3/2-	0-	3/2-	1+	3	\rightarrow	3.2-	\rightarrow	32-	(3-)	32-	(2+,3+)	(3/2-)
JU	EC	EC	EC	EC	EC	EC	69,165	ECS	39.882	9		β		9	
D	Zn62	Zn63	Zn64	Zn65	Zn66	Zn67	Zn68	Zn69	Zn70	Zn71	Zn72	Zn73	Zn74	Zn75	Zn76
	0+	3/2-	0+	5/2-	0+	52-	<u></u> !:	-	0+	→ I	0-	(1/2)	0+	(7/2+)	0+
0	EC	EC	45.6	EC	27.9	41	18.8	β.	0.6	β.	8	β	\$		9
P	Cuól	Cu62	Cu63	Cu64	Cu65	Cu66	Cu67	Cu68	Cu69	Cu70	Cu71	Cu72	Cu73	Cu74	Cu75
	3/2-	1+	3/2	\rightarrow	3/3	→ ·	3/2-	1+	3/2-	(1+)	(3/2-)	(1+)		(1+,5+)	1
	EC	EC	69.17	ECR	30.55	8	9	e *	8	9		8	8	9	8.
	Ni60	Ni61	Ni62	N103	Ni64	N105	N100	N107	N168	N169	N170	N171	N172	N173	N174
	0+	3/2-	6+	•	0+	2	0+	(1/2-)	0+	11.4 s	0+	130 1	0+	0301	0+
	26.223	1.140	3.634	9	0,926	8	9	9	8	9		8	8	9	9

Neutron number

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B²FH

Synthesis of the Elements in Stars*

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> "It is the stars, The stars above us, govern our conditions"; (King Lear, Act IV, Scene 3)

> > but perhaps

"The fault, dear Brutus, is not in our stars, But in ourselves," (Julius Caesar, Act I, Scene 2)

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										1.25							

NEUTRON STARS

What happened to the iron core after the supernova?





NEUTRON STAR

N N N N N Ν N N N N N N N N N N N N N N N N N N N Ν N Ν N N N N N Ν N N N N N N Ν N N N N N N Ν N N N Ν N N N Ν Ν Ν N N N N N Ν N Ν N N N N N N N N N N

neutron star

Solar-mass white dwarf

Earth

Model of a Neutron Star

Mass ~1.5 times the Sun

Solid crust
 ~1 mile thick

Diameter
 ~ 12 miles

Heavy liquid interior Mostly neutrons, with other particles



What keeps the neutron star from collapsing? Ν Ν Ν Ν Ν Ν Ν Ν Ν Ν

NEUTRON DEGENERACY PRESSURE Neutrons have a limit to how tightly they can be packed together

Ν

Ν

Chandrasekhar Limit for neutron stars

M < 3.0 M.